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Stoichiometry
Problems 1

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Basic

Introduction,

Mole to Mole,

Grams to Grams,

Mole Ratio

Practice

Problems Step by

Step

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Practice

Problems | How

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Massive

Creatures: Crash

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Test Review

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Practice Quiz

(Honors

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2018

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Stoichiometry 2:

moles and grams

~~Honors Chem 325:~~

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Lecture # 2

Stoichiometry

Made Easy:

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Magic Number

Method

Chemistry Final

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Moles to Grams

Stoichiometry

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*Stoichiometry
Tutorial: Step
by Step Video +
review problems
explained |
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Reactant \u0026
Excess Reactant
Stoichiometry*

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1u0026 Moles

Stoichiometry
with Mass:

Stoichiometry

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Worksheet # 3

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~~Stoichiometry 3:~~

~~grams to grams~~

~~Honors~~

~~Chemistry~~

~~Stoichiometry 4:~~

~~moles and liters~~

Intro To

Stoichiometry |

AP/Honors

Chemistry

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Chemistry-

Stoichiometry 5:

Summary

Flowchart Honors

Chemistry,

5/4/2020,

Stoichiometry

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Stoichiometry

Worksheet #1

Answers.

Stoichiometry

Worksheet #1

Answers 1. Given

the following

equation: $2 \text{C}_4\text{H}_{10} + 13 \text{O}_2 \rightarrow$

$8 \text{CO}_2 + 10 \text{H}_2\text{O}$, show what

the following

molar ratios

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should be a. C
4H 10 / O 2 b. O
2 / CO 2 c. O 2
/ H 20 d. C 4H
10 / CO 2 e. C
4H 10 / H 20 2.

Given the
following
equation: $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$
a.

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Practice 1

Answers

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Extra

Stoichiometry

Problems 1.

Silver nitrate

reacts with

barium chloride

to form silver

chloride and

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barium nitrate.

a. Write and balance the chemical

equation. 2AgNO_3

$+ \text{BaCl}_2 \rightarrow 2 \text{AgCl}$

$+ \text{Ba}(\text{NO}_3)_2$

b. If 39.02

grams of barium chloride are

reacted in an

excess of silver nitrate, how

many

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Answers . . . 1

Dr. Rodriguez-

Reyes Chemistry

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problems I.

Answer the

questions for

the following

reaction: Na (s)

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(s) 1. How many moles of Na are needed to react with 12.50 mole Cl_2 ? 2. How many moles of Na are needed to produce 56.79 mole of NaCl? 3. How many moles of Cl_2 are needed to react with 33.50 mole

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Na? 4. Honors

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2018

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Dr Rodriguez-

Reyes ...

Homework—Solving

Stoichiometry

Problems Name

_____ ANSWERS

_____ If the

statement is

true, write

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“true”. If it is false, change the underlined word or words to make it true.

Write your answer on the line provided.

 TRUE 1.

The major types of stoichiometry problems are mass-mass, mass-volume, and

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Volume-volume.

Chemistry

Homework—Solving

Stoichiometry

Problems 1

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Practice

Worksheet –

Stoichiometry.

1. How many moles of oxygen are consumed when 96.7 moles of hydrogen

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Sulfide gas are burned, producing sulfur dioxide and water vapor in the process? 2. If 3.70×10^{23} molecules of oxygen react with excess benzene (C_6H_6), how many grams of water can be produced? 3.

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Problems 1.

Silver nitrate

reacts with

barium chloride

to form silver

chloride and

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a. Write and balance the chemical

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$+ \text{BaCl}_2 \rightarrow 2 \text{AgCl}$

$+ \text{Ba}(\text{NO}_3)_2$

b. If 39.02

grams of barium chloride are

reacted in an

excess of silver nitrate, how

many

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Problems 1

HW4 Solutions-Mo
larity-

Stoichiometry WS
1-14 Answers

Page 1 Page 2

HW5 Activity

5-8: 1-8 Answers

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Chemistry Final
Review . Honors
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Stoichiometry
Assignments.

Acids and Bases

TEST Wed June 7

HW1 (5/30)

Definitions -
handed in ...

Bader, Mr. K. -
Science / Honors
Chem Homework
The

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Stoichiometry of
Alka-Seltzer.

This lab will
count as your

Honors Project.

You will submit
a typed, formal
lab report,

including all
pre-lab and post-
lab assignments.

It will count as
a formal grade
worth 100

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points. Alka-Seltzer is one of the world's best-known antacids. Its main function is to absorb excess stomach acid (HCl).

The
Stoichiometry of
Alka Seltzer
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homework

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and ask

questions when
you need
clarification.

... Chapter 12

SG 12.1

Introduction to
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Mr.

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P.S.#2. A

student performs
a double
replacement
reaction by
mixing 500.0 ml
of a 0.228 M
solution of
lithium

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carbonate with
370.0 mL of a
0.352 M solution
of iron(III)
chloride. The
student collects
the precipitate,
and finds that
9.98 grams of
precipitate are
recovered. 1.
Write the net
ionic equation.

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is designed for
students who
have
demonstrated
strong ability
in previous
science courses
Unit 8
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demanding
course, the main
topics--which
include atomic
theory, nuclear
chemistry,
periodicity,
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reactions,
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gases,
solutions,
reaction

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kinetics,
equilibrium,
acid-base
theory, oxidatio
n-reduction, and
...

Unit 8

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Answer Key

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Problems 1) 10+

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CO CO >>> Co.

a). 2.00g of
carbon monoxide
reacted with

duodine

pentonde,
calculate the
theoretical

yield of 1 b).

If 3.179 of t,
was

experimentally
(actually)

produced,

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PDF Answers

calculate the
percent yield of
1, 2). CH_4 , NH_3 ,
 $+ \text{H}_2$ CH_2N $+ \text{H}_2\text{O}$

a). Problems 1

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Problems 1) 10+

CO CO $\gg \text{Co}$...

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Measurement

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Structure

Answers 02b

Atomic Theory &

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Summary Answers

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Symbols Answers

03b Inorganic

Nomenclature I

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Using the mole

ratio

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(stoichiometry)
of 1 mol NO : 3
mol NO₂ (or it
takes 3 mol NO₂
to make 1 mol
NO)... we can
set up the
following
relationship
(using
dimensional
analysis): 1.2
moles NO₂ x 1
mole NO / 3

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moles $\text{NO}_2 = 0.4$

moles NO formed.

82). Write a

correctly

balanced

equation for the

reaction taking

place: $2\text{NO}(\text{g}) +$

$\text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2$

(g)

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