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tration of the acid? A 50.0-mL sample of 0.0100M is neutralized by 45.6 ml. of HBr. What is the molarity Of the acid? 9. Whatis its [H+]? 10. What is its pH? 11. What is its pOH? A solution has a [1-14] of 5.6 X 10 12. What is its [OH-]? Supplemental Problems 29 -6M. Chemistry: Matter and Change Chapter 19

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