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Physical Changes

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Amazing Experiments
with Water

Pure Substances and
Mixtures!

(Classification of
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Chemical Changes

Understanding Atoms,
elements, and

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molecules Part #1

(9min)

Basic Chemistry

Series - 1.2 What is

matter? - HD

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Liquids, Gases

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Pure Substances and

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Introduction to

Chemistry 1.1 The

Stories of Two

Chemicals Ozone

Layer, atmosphere,

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ozone formation,
chlorofluorocarbons,
CFC ' s 1.2 Chemistry
and Matter Chemistry
Central Science

Chapter Summaries
– Chemistry Matter
and Change

Chapter 1 Section 2

MATTER and Its
Properties Matter –
Anything that has
MASS and VOLUME.

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ATOMS and **MOLECULES** the basic building blocks of matter. **ATOM**-- the smallest unit of an element that still maintains the properties that element. **ELEMENT**-- a pure substance made up of only one type of atom.

MATTER and
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CHANGE Chapter 1

Section 1

Chemistry: Matter
and Change Chapter

6 Study Guide for

Content Mastery .

Name Date CHAPTER

Class STUDY GUIDE

FOR CONTENT

MASTERY Section 6.3

Periodic Trends In

your textbook, read

about atomic radius

and ionic radius.

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Circle the letter of the choice that best completes the statement or answers the question. 1.

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CHAPTER 3

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SOLUTIONS MANUAL

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Chapter 3 35 Section

3.1 Properties of

Matter pages 70–75

Problem-Solving Lab

1. Explain why the
flow of a compressed
gas must be

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controlled for
practical and safe use.

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and Changes Matter—
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Chapter 1 VIDEODISC

Cosmic Chemistry

Greenhouse

Effect, Movie Ozone

Studies, Movie Nylon,

Movie Dr. Jekyll and

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Safety, Movie CD-ROM

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Chemistry
Chemistry: Matter
and Change Teacher

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Guide and Answers 7

Study Guide - Chapter

6 – The Periodic

Table and Periodic

Law Section 6.1

Development of the

Modern Periodic

Table 1. octaves 2.

eight 3. nine 4.

accepted 5. Dmitri

Mendeleev 6. atomic

mass 7. elements 8.

atomic number 9.

Henry Moseley 10.

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protons 11. periodic
law 12. properties 13.
14 ...

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Ch 6 Study Guide

answers

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Assessment

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Chapter 5 25

Electrons in Atoms

Reviewing

Vocabulary Match the

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definition in Column
A with the term in
Column B. Column A
Column B 1. The set
of frequencies of the
electromagnetic
waves emitted by the
atoms of an element
2. The minimum
amount of energy
that can be lost or
gained by ...

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that can exchange

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extensive property.
intensive property.
extensive properties
(examples) intensive
property (examples) a

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Property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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change chapter 2
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...

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Math Skills And

Transparency
Change Chapter

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Chapter 11 31
Answer Key

number of moles of
known substance C H
C H C H O O O CO CO
CO H O H O H O

number of moles of
unknown substance O
CO H O C H CO H O C
H O H O C H O CO

use mole ratio 5 mol

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0.21 mol C₃H₈

mol CO₂ 1 mol C₃H₈

8.4 mol H₂O 1 mol

C₃H₈ 1 mol C₃H₈

8.5 mol O₂

MATH SKILLS

TRANSPARENCY

MASTER 16

A “chemical change” is a change that produces matter with a different composition than the

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original matter.

Chemical Change A change in which one or more substances are converted into different substances.

Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different properties than their component elements.

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Containing 52 tested
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sequence and
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comprehensive

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Chemistry course of study designed for a first-year high school chemistry curriculum.

The program incorporates features for strong math support and problem-solving development. The content has been reviewed for accuracy and significant enhancements have been made to provide

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and teacher-driven
technology support.
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This student
companion is a
supplement to
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Matter, and Change,
4th edition with CD-
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guided reading

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Nuclear chemistry.

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Appendices. Glossary.

Answers. Illustration

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Answer Key

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