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Chapter 1 Section 2 MATTER and Its Properties Matter – Anything
that has MASS and VOLUME. ATOMS and MOLECULES the basic

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building blocks of matter. **ATOM**-- the smallest unit of an element that still maintains the properties that element. **ELEMENT** – a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1 Section 1

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CONTENT MASTERY Section 6.3 Periodic Trends In your textbook,
read about atomic radius and ionic radius. Circle the letter of the
choice that best completes the statement or answers the question. 1.

Atomic radii cannot be measured ...

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of Matter pages 70 – 75 Problem-Solving Lab 1. Explain why the flow
of a compressed gas must be controlled for practical and safe use.

Matter—Properties and Changes Matter—Properties and Changes
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Chapter 5 25 Electrons in Atoms Reviewing Vocabulary Match the definition in Column A with the term in Column B. Column A
Column B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

Chapter Assessment

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intensive property. extensive properties (examples) intensive property (examples) a property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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Chapter 11 31 number of moles of known substance C H C H C H O
O O CO CO CO H O H O H O number of moles of unknown
substance O CO H O C H CO H O C H O H O C H O CO use mole
ratio 5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O
1 mol C 3 H 8 1 mol C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY MASTER 16

A “ chemical change ” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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