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An Introduction to Thermal Physics

1. DIESEL ENGINES 2. As in the Otto cycle, the red and blue

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curves are adiabats, so no heat is exchanged during these steps. The efficiency of the Diesel cycle is most easily worked out by calculating the heat input  $Q$  along the yellow step 2 to 3 and the heat expelled during the green step 4 to 1. The efficiency is then  $e = W/Q$ .

DIESEL ENGINES An Introduction to Thermal Physics in Schroeder),  $\Delta H_{H_2O} = 285.83 \text{ kJ mol}^{-1}$  and  $\Delta H_{H_2O} = 890.36 \text{ kJ mol}^{-1}$  (12) This time, the final volume is  $1/3$  of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work  $PdV = RT \ln n$  (13)  $= 8.31 \text{ J K}^{-1} (298 \text{ K})(2)$  (14)  $= 4.953 \text{ kJ}$  (15)

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