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TEACHING TRANSPARENCY
WORKSHEET Ionic Bonds Use with
Chapter 7, Section 7.2 1. How many
valence electrons does a neutral
magnesium (Mg) atom have? 2. What is
the charge on a magnesium ion? What
does magnesium have to do to form such
an ion, and why does it tend to do so? 3.

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How many valence electrons does a single neutral chlorine (Cl) atom have? 4.

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Chemistry: Matter and Change Chapter 5

Lesson 11 Name: Date: Hour: Information:

Energy Levels and Sublevels ...

Considering your answer to question 5,
how many electrons can fit in the entire 5th
energy level? (Answer is a half AC5a)

5. Write the notation for an
electron spinning clockwise in a p sublevel

...

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WORKSHEET The Activity Series Use
with Chapter 9, Section 9.2 1. For each of
the following pairs of elements, underline

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the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

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WORKSHEET The Activity Series
Teaching Transparency Worksheets
Chemistry Answers Chapter 19 Author:
s2.kora.com-2020-11-30T00:00:00+00:01
Subject: Teaching Transparency
Worksheets Chemistry Answers Chapter
19 Keywords: teaching, transparency,
worksheets, chemistry, answers, chapter,
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Chemistry: Matter and Change 0 Chapter
8 . 1.7, the CF bond is polar covalent. (b)

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!ENCI"P = ENCI " ENP. = 3.0" 2.1

!ENCI"P = 0.9. Since $0.5 < \text{EN. A.} < 1.7$,
the PCI bond is polar covalent. (c)

!ENCI"Li Section 8.5 Electronegativity
and Polarity 265 Compare and contrast
polar . that is in excess.

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WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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